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# Parallel Kinetic Resolution of Acyclic $\gamma$ -Amino- $\alpha$ , $\beta$ -unsaturated Esters: Application to the Asymmetric Synthesis of 4-Aminopyrrolidin-2-ones

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Conjugate addition of a 50:50 pseudoenantiomeric mixture of lithium (*R*)-*N*-benzyl-*N*-( $\alpha$ -methylbenzyl)amide and lithium (*S*)-*N*-3,4-dimeth-oxybenzyl-*N*-( $\alpha$ -methylbenzyl)amide to a range of racemic acyclic  $\gamma$ -amino- $\alpha$ , $\beta$ -unsaturated esters (derived from the corresponding  $\alpha$ -amino acids) effects their efficient parallel kinetic resolution, allowing the preparation of enantiopure  $\beta$ , $\gamma$ -diamino esters. The  $\beta$ , $\gamma$ -diamino ester products of these reactions are readily converted into the corresponding substituted 4-aminopyrrolidin-2-ones via *N*-debenzylation and cyclization.

Kinetic resolution is a venerable concept within organic chemistry. It was first observed by Marckwald and McKenzie in 1899, during the esterification of racemic

(1) Marckwald, W.; McKenzie, A. Ber. Dtsch. Chem. Ges. 1899, 32, 2130.

mandelic acid with (-)-menthol.<sup>1</sup> Such is the importance of kinetic resolution that over 100 years later it, together with dynamic kinetic resolution (DKR)<sup>2</sup> and parallel kinetic resolution (PKR),<sup>3</sup> is employed widely in the preparation of enantiomerically pure materials on both laboratory and industrial scales. We have previously developed the PKR of a range of chiral 3- and 5-substituted cyclopent-1-enecarboxylates, and 6-substituted cyclohex-1-enecarboxylates using a 50:50 pseudoenantiomeric mixture of enantiopure lithium amides.<sup>4</sup> This protocol has thus far been limited to these cyclic  $\alpha$ ,  $\beta$ -unsaturated esters due to the requirement for high levels of substrate control. In this manuscript we report the efficient PKR of racemic acyclic *N*-protected  $\gamma$ -amino- $\alpha$ . $\beta$ -unsaturated esters (which exhibit very high levels of substrate control) using a 50:50 pseudoenantiomeric mixture of lithium (R)-N-benzyl-N-( $\alpha$ -methylbenzyl)amide and lithium (S)-N-3,4-dimethoxybenzyl-N-( $\alpha$ -methylbenzyl)amide. The enantiopure

<sup>(2)</sup> Pellissier, H. Tetrahedron 2003, 59, 8291. Pellissier, H. Tetrahedron 2008, 64, 1563.

<sup>(3)</sup> Eames, J. Angew. Chem., Int. Ed. 2000, 39, 885. Dehli, J. R.; Gotor, V. Chem. Soc. Rev. 2002, 31, 365.

<sup>(4)</sup> Davies, S. G.; Díez, D.; El Hammouni, M. M.; Garner, A. C.; Garrido, N. M.; Long, M. J. C.; Morrison, R. M.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Chem. Commun. 2003, 2410. Davies, S. G.; Garner, A. C.; Long, M. J. C.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Org. Biomol. Chem. 2004, 2, 3355. Davies, S. G.; Garner, A. C.; Long, M. J. C.; Morrison, R. M.; Roberts, P. M.; Savory, E. D.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Org. Biomol. Chem. 2005, 3, 2762. Aye, Y.; Davies, S. G.; Garner, A. C.; Roberts, P. M.; Smith, A. D.; Thomson, J. E. Org. Biomol. Chem. 2008, 6, 2195. Abraham, E.; Davies, S. G.; Gocherty, A. J.; Ling, K. B.; Roberts, P. M.; Russell, A. J.; Thomson, J. E.; Toms, S. M. Tetrahedron: Asymmetry 2008, 19, 1356. Davies, S. G.; Durbin, M. J.; Hartman, S. J. S.; Matsuno, A.; Roberts, P. M.; Russell, A. J.; Smith, A. D.; Thomson, J. E.; Toms, S. M. Tetrahedron: Asymmetry 2008, 19, 1356.

 $\beta$ , $\gamma$ -diamino ester products of these reactions are valuable building blocks for further elaboration, as demonstrated by their facile conversion to the corresponding substituted 4aminopyrrolidin-2-ones.

A range of racemic acyclic *N*-protected  $\gamma$ -amino- $\alpha,\beta$ -unsaturated esters was prepared from the corresponding racemic  $\alpha$ -amino acids (±)-1-7 using a modification of the procedure reported by Reetz and co-workers.<sup>5</sup> Exhaustive benzylation of (±)-1-7 was achieved upon treatment with BnBr in boiling aq K<sub>2</sub>CO<sub>3</sub> to give (±)-8-14 and was followed by reduction with LiAlH<sub>4</sub> to give the corresponding *N*,*N*-dibenzyl protected  $\alpha$ -amino alcohols (±)-15-21. Swern oxidation of  $\alpha$ -amino alcohols (±)-15-21 and olefination of the resultant aldehydes then gave the desired  $\alpha,\beta$ -unsaturated esters (±)-22-28 in 18-57% overall yield (Scheme 1).

### Scheme 1



When investigating PKR,<sup>4</sup> we have promulgated that it is prudent to follow a strategy of first investigating the levels of substrate control offered by the chiral  $\alpha,\beta$ unsaturated ester upon conjugate addition of an achiral lithium amide, viz. lithium N-benzyl-N-isopropylamide 29. The levels of enantiorecognition between the chiral  $\alpha_{\beta}$ unsaturated ester (substrate) and lithium N-benzyl-N-( $\alpha$ methylbenzyl)amide 30 (chiral reagent) are then evaluated by investigation of their mutual kinetic resolution (MKR), i.e., addition of racemic lithium amide  $(\pm)$ -30 to racemic  $\alpha,\beta$ -unsaturated ester. This approach eliminates any complicating effects of mass action and allows the maximum levels of enantiodiscrimination (as quantified by the factor, E)<sup>6</sup> to be very simply determined by analysis of the product distribution by <sup>1</sup>H NMR spectroscopy. Finally, having identified those substrates that undergo efficient MKR upon addition of racemic lithium amide  $(\pm)$ -30, their PKR employing a 50:50 pseudoenantiomeric mixture of enantiopure lithium N-benzyl-N-( $\alpha$ -methylbenzyl) amide 30 and enantiopure lithium N-3,4-dimethoxybenzyl-N-( $\alpha$ -methylbenzyl)amide **31** may be performed. We therefore adopted this approach to investigate the potential of racemic  $\gamma$ -amino- $\alpha$ , $\beta$ -unsaturated esters ( $\pm$ )-**22–28** as substrates for our PKR protocol.

Addition of 1.6 equiv of lithium amide 29 to  $(\pm)$ -22–26 resulted in >95% conversion to the corresponding  $\beta$ ,  $\gamma$ -diamino esters ( $\pm$ )-32-36 ( $\geq$  88:12 dr in all cases), indicating high levels of substrate control. Chromatographic purification allowed isolation of  $\beta$ ,  $\gamma$ -diamino esters ( $\pm$ )-32–36 in 60-86% yield and in  $\geq$  97:3 dr (Scheme 2). The relative 3,4syn-configuration within  $(\pm)$ -34 (R = Bn) was unambiguously established by single crystal X-ray diffraction analysis of the corresponding hydrochloride salt  $(\pm)$ -34•HCl. The relative 3,4-*svn*-configurations within  $(\pm)$ -32,  $(\pm)$ -33,  $(\pm)$ -35, and  $(\pm)$ -36 were assigned by analogy.<sup>7</sup> Meanwhile,  $(\pm)$ -28 proved recalcitrant to addition of lithium amide 29, even over extended reaction times and when the amount of lithium amide was increased from 1.6 to 10 equiv. Addition to  $(\pm)$ -27 proceeded with low levels of substrate control to give a 27:73 mixture of 3,4-svn-37:3,4-anti-43, which were isolated in 15 and 48% yield as single diastereoisomers (>99:1 dr). The relative 3,4anti-configuration within 43 was unambiguously established by single crystal X-ray diffraction analysis, which therefore allowed the relative 3,4-syn-configuration within 37 to be assigned unambiguously (Scheme 2).



Previous investigations concerning conjugate addition of a range of nucleophiles to  $\alpha,\beta$ -unsaturated carbonyl compounds with a stereocenter at the  $\gamma$ -position often invoke a modified Felkin–Anh model to rationalize the

<sup>(5)</sup> Reetz, M. T.; Röhrig, D. Angew. Chem., Int. Ed. Engl. 1989, 28, 1706.

<sup>(6)</sup> Horeau, A. *Tetrahedron* **1975**, *31*, 1307.

<sup>(7)</sup> Comparison with the substrate control elicited in the MKR of  $(\pm)$ -22–26 with lithium amide  $(\pm)$ -30, as well as in the PKR of  $(\pm)$ -22–26 with lithium amides (*R*)-30 and (*S*)-31, allows these configurational assignments to be made confidently. In addition, the relative configurations within  $\beta$ , y-diamino esters  $(\pm)$ -36 and  $(\pm)$ -42 [and  $(3S,4R,\alpha R)$ -48] were subsequently unambiguously established by single crystal X-ray diffraction analysis of a cyclic derivative; see the Supporting Information for full experimental details.

<sup>(8)</sup> Yamamoto, Y.; Chounan, Y.; Nishii, S.; Ibuka, T.; Kitahara, H. J. Am. Chem. Soc. **1992**, 114, 7652. Kireev, A. S.; Manpadi, M.; Kornienko, A. J. Org. Chem. **2006**, 71, 2630.

observed diastereoselectivity.8 However, one other simplistic model (not dissimilar to a Felkin-Anh model) which is able to rationalize successfully the experimental data in this case uses insight obtained from single crystal X-ray diffraction analysis of  $\alpha,\beta$ -unsaturated ester (±)-22. This revealed a solid state conformation in which the C(4)hydrogen atom lies almost perpendicular to the plane of the  $\alpha$ . $\beta$ -unsaturated system, with the bulky C(4)-N,N-dibenzylamino substituent occupying the less hindered "outside" position and the C(4)-methyl group in the more hindered "inside" position (Figure 1). Conjugate addition of lithium amide 29 to  $(\pm)$ -22 in this conformation would be predicted to occur on the least hindered face past the "small" hydrogen substituent to give  $(\pm)$ -3,4-syn-32, as observed experimentally. A similar analysis applied to  $\alpha_{\beta}$ -unsaturated esters (±)-23-26 (R = Et, Bn, <sup>i</sup>Bu, CH<sub>2</sub>OBn) would also successfully rationalize the observed substrate diastereofacial control, leading to  $(\pm)$ -3.4-syn-33-36. However, increased steric bulk of the C(4)-substituents in  $\alpha,\beta$ -unsaturated esters (±)-27 and (±)-28 would serve to disfavor analogous conformations, thereby providing a rationale for their differing behavior. Presumably, the very large steric congestion around C(4) in the case of  $(\pm)$ -28  $(\mathbf{R} = {}^{i}\mathbf{P}\mathbf{r})$  precludes addition of the sterically demanding lithium amide to C(3) completely.



**Figure 1.** Chem 3D representation of the single crystal X-ray diffraction structure of  $(\pm)$ -**22** [(*S*)-enantiomer depicted; selected H atoms are omitted for clarity], and Newman projection along the C(3)–C(4) bond.

The extent of enantiorecognition between  $\alpha,\beta$ -unsaturated esters  $(\pm)$ -22-26 (which offered high levels of substrate control) and lithium amide  $(\pm)$ -30 was next investigated, with high levels of enantiorecognition being expected. Indeed, addition of 1.6 equiv of lithium amide  $(\pm)$ -30 to  $(\pm)$ -22–26 gave, in each case, essentially a single diastereoisomeric product 44-48 in  $\geq 95:5$  dr, indicating very high levels of enantiorecognition between substrate and reagent, and consistent with  $E \ge 19^6$  in each case. Purification facilitated isolation of diastereoisomerically pure (>99:1 dr) samples of  $(\pm)$ -44–48. The relative  $(3RS, 4RS, \alpha SR)$ -configurations within  $(\pm)$ -45 and  $(\pm)$ -47 were unambiguously established by single crystal X-ray diffraction analyses, and therefore the relative configurations within  $(\pm)$ -44,  $(\pm)$ -46, and  $(\pm)$ -48 were assigned by analogy. It is notable that the relative configurations of the C(3)- and C( $\alpha$ )-stereogenic centers within both (±)-45 and

Scheme 3



<sup>*a*</sup> 70% of a sample of **46** in 95:5 dr was also isolated.

(±)-47 [and, hence, (±)-44, (±)-46, and (±)-48] are in accordance with that predicted by the transition state mnemonic developed by us to rationalize the exceptional facial bias of this class of lithium amide.<sup>9</sup> This reagent control, when combined with that of the  $\alpha,\beta$ -unsaturated ester (substrate control: production of the 3,4-*syn*-diastereoisomer favored), results in very highly selective reactions. These results suggest that  $\alpha,\beta$ -unsaturated esters (±)-22–26 are viable substrates for our PKR protocol<sup>4</sup> (Scheme 3).

The PKR of  $\alpha$ . $\beta$ -unsaturated esters (±)-22-26 using a 50:50 pseudoenantiomeric mixture of lithium amides (R)-30 (2 equiv) and (S)-31 (2 equiv) was next investigated. These reactions produced, in each case, a 50:50 mixture of the corresponding ( $\alpha R$ )-adducts 44–48 in  $\geq$  95:5 dr and the ( $\alpha$ S)-adducts **49–53** in  $\geq$ 95:5 dr. Facile separation and purification via flash column chromatography allowed isolation of  $(\alpha R)$ -44-48 in > 99:1 dr and 31-42% yield and  $(\alpha S)$ -49–53 in > 99:1 dr and 40–43% yield. In each case, the product of addition of lithium amide (R)-30 was spectroscopically identical to the major diastereoisomer formed in the corresponding MKR reaction. Additionally, the relative configuration within  $\beta$ ,  $\gamma$ -diamino ester 49 was unambiguously established via single crystal X-ray diffraction analysis, with the absolute  $(3R, 4R, \alpha S)$ -configuration being assigned from the known (S)-configuration of the  $\alpha$ -methylbenzyl stereocenter. Given the pseudoenantiomeric nature of lithium amides (R)-30 and (S)-31, this analysis

<sup>(9)</sup> Costello, J. F.; Davies, S. G.; Ichihara, O. *Tetrahedron: Asymmetry* **1994**, *5*, 1999. For a review, see: Davies, S. G.; Smith, A. D.; Price, P. D. *Tetrahedron: Asymmetry* **2005**, *16*, 2833.

<sup>(10)</sup> Hoang, C. T.; Bouillère, F.; Johannesen, S.; Zulauf, A.; Panel, C.; Pouilhès, A.; Gori, D.; Alezra, V.; Kouklovsky, C. *J. Org. Chem.* **2009**, *74*, 4177.

<sup>(11)</sup> In the case of  $\beta$ , $\gamma$ -diamino ester **48**, the yield of the corresponding 4-aminopyrrolidin-2-one **68** (60% isolated yield) was somewhat compromised by the formation of methyl (3*S*,4*R*)-3,4-diacetamido-5-hydro-xypentanoate **69** (25% isolated yield). This presumably arises from competing lactone formation (rather than lactam formation) from **58** under the reaction conditions, followed by methanolysis upon workup of the acetylation procedure.

#### Scheme 4



also allows the assigned relative  $(3RS,4RS,\alpha SR)$ -configuration within racemic **44** to be unambiguously confirmed, with the absolute  $(3S,4S,\alpha R)$ -configuration within enantiopure **44** following from the known (*R*)-configuration of the  $\alpha$ -methylbenzyl stereocenter. By similar reasoning, given the known relative configurations within racemic **45** and **47**, the absolute  $(3S,4S,\alpha R)$ -configurations within enantiopure **45** and **47** may be assigned from the known (*R*)-configuration of the  $\alpha$ -methylbenzyl stereocenter. Hence, the absolute  $(3R,4R,\alpha S)$ -configurations within **50** and **52** can be unambiguously assigned. The absolute  $(3R,4R,\alpha S)$ -configurations within **51** and **53** were assigned by analogy (Scheme 4).

With a range of enantiopure  $\beta$ , $\gamma$ -diamino esters in hand, their synthetic utility was demonstrated by elaboration of **44–48** to the corresponding 5-substituted 4-aminopyrrolidin-2-ones. Hydrogenolytic *N*-debenzylation of **44–48** was followed by acid-promoted cyclization<sup>10</sup> to the corresponding 4-aminopyrrolidin-2-ones **59–63**, which were isolated as their acetate derivatives **64–68** in 60–78% yield over three steps.<sup>11</sup> The absolute configurations within **64–68** were assigned from the known absolute Scheme 5



<sup>*a*</sup> Yield over 3 steps. <sup>*b*</sup> For 48, R' = Bn; for 58 and 63, R' = H; for 68, R' = Ac.

configurations of the precursor  $\beta$ , $\gamma$ -diamino esters **44**–**48**; <sup>1</sup>H NMR NOE analyses of **64**–**68** were also supportive of a relative 4,5-*syn*-configuration (Scheme 5).

In conclusion, conjugate addition of a 50:50 pseudoenantiomeric mixture of lithium (*R*)-*N*-benzyl-*N*-( $\alpha$ -methylbenzyl)amide and lithium (*S*)-*N*-3,4-dimethoxybenzyl-*N*-( $\alpha$ -methylbenzyl)amide to a range of racemic acyclic  $\gamma$ -amino- $\alpha$ , $\beta$ -unsaturated esters (derived from the corresponding  $\alpha$ -amino acids) effects their efficient parallel kinetic resolution, allowing the preparation of enantiopure  $\beta$ , $\gamma$ -diamino esters. The  $\beta$ , $\gamma$ -diamino ester products of these reactions are readily converted into the corresponding substituted 4-aminopyrrolidin-2-ones via *N*-debenzylation and cyclization. Further applications of this methodology are currently under investigation within our laboratory.

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**Supporting Information Available.** Experimental procedures, characterization data, copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra, and crystallographic information files (for structures CCDC 852571–852577). This material is available free of charge via the Internet at http://pubs.acs.org.