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# Parallel Kinetic Resolution of Acyclic  $γ$ -Amino-α, $β$ -unsaturated Esters: Application to the Asymmetric Synthesis of 4-Aminopyrrolidin-2-ones

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Conjugate addition of a 50:50 pseudoenantiomeric mixture of lithium  $(R)$ -M-benzyl-N-( $\alpha$ -methylbenzyl)amide and lithium (S)-N-3,4-dimethoxybenzyl-N-(α-methylbenzyl)amide to a range of racemic acyclic  $\gamma$ -amino-α,β-unsaturated esters (derived from the corresponding  $\alpha$ -amino acids) effects their efficient parallel kinetic resolution, allowing the preparation of enantiopure  $β,γ$ -diamino esters. The  $β,γ$ -diamino ester products of these reactions are readily converted into the corresponding substituted 4-aminopyrrolidin-2-ones via N-debenzylation and cyclization.

Kinetic resolution is a venerable concept within organic chemistry. It was first observed by Marckwald and McKenzie in 1899, during the esterification of racemic mandelic acid with  $(-)$ -menthol.<sup>1</sup> Such is the importance of kinetic resolution that over 100 years later it, together with dynamic kinetic resolution  $(DKR)^2$  and parallel kinetic resolution  $(PKR)$ ,<sup>3</sup> is employed widely in the preparation of enantiomerically pure materials on both laboratory and industrial scales. We have previously developed the PKR of a range of chiral 3- and 5-substituted cyclopent-1-enecarboxylates, and 6-substituted cyclohex-1-enecarboxylates using a 50:50 pseudoenantiomeric mixture of enantiopure lithium amides.<sup>4</sup> This protocol has thus far been limited to these cyclic  $\alpha$ , $\beta$ -unsaturated esters due to the requirement for high levels of substrate control. In this manuscript we report the efficient PKR of racemic acyclic N-protected γ-amino-α, $\beta$ -unsaturated esters (which exhibit very high levels of substrate control) using a 50:50 pseudoenantiomeric mixture of lithium  $(R)$ -N-benzyl-N- $(\alpha$ -methylbenzyl)amide and lithium  $(S)-N-3$ ,4-dimethoxybenzyl- $N-(\alpha$ -methylbenzyl)amide. The enantiopure

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<sup>(2)</sup> Pellissier, H. Tetrahedron 2003, 59, 8291. Pellissier, H. Tetrahedron 2008, 64, 1563.

<sup>(3)</sup> Eames, J. Angew. Chem., Int. Ed. 2000, 39, 885. Dehli, J. R.; Gotor, V. Chem. Soc. Rev. 2002, 31, 365.

<sup>(4)</sup> Davies, S. G.; Díez, D.; El Hammouni, M. M.; Garner, A. C.; Garrido, N. M.; Long, M. J. C.; Morrison, R. M.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Chem. Commun. 2003, 2410. Davies, S. G.; Garner, A. C.; Long, M. J. C.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Org. Biomol. Chem. 2004, 2, 3355. Davies, S. G.; Garner, A. C.; Long, M. J. C.; Morrison, R. M.; Roberts, P. M.; Savory, E. D.; Smith, A. D.; Sweet, M. J.; Withey, J. M. Org. Biomol. Chem. 2005, 3, 2762. Aye, Y.; Davies, S. G.; Garner, A. C.; Roberts, P. M.; Smith, A. D.; Thomson, J. E. Org. Biomol. Chem. 2008, 6, 2195. Abraham, E.; Davies, S. G.; Docherty, A. J.; Ling, K. B.; Roberts, P. M.; Russell, A. J.; Thomson, J. E.; Toms, S. M. Tetrahedron: Asymmetry 2008, 19, 1356. Davies, S. G.; Durbin, M. J.; Hartman, S. J. S.; Matsuno, A.; Roberts, P. M.; Russell, A. J.; Smith, A. D.; Thomson, J. E.; Toms, S. M. Tetrahedron: Asymmetry 2008, 19, 2870.

 $\beta$ , *γ*-diamino ester products of these reactions are valuable building blocks for further elaboration, as demonstrated by their facile conversion to the corresponding substituted 4 aminopyrrolidin-2-ones.

A range of racemic acyclic N-protected  $\gamma$ -amino- $\alpha$ , $\beta$ -unsaturated esters was prepared from the corresponding racemic  $\alpha$ -amino acids ( $\pm$ )-1-7 using a modification of the procedure reported by Reetz and co-workers.<sup>5</sup> Exhaustive benzylation of  $(\pm)$ -1-7 was achieved upon treatment with BnBr in boiling aq  $K_2CO_3$  to give ( $\pm$ )-8-14 and was followed by reduction with  $LiAlH<sub>4</sub>$  to give the corresponding N,N-dibenzyl protected  $\alpha$ -amino alcohols ( $\pm$ )-15–21. Swern oxidation of  $\alpha$ -amino alcohols ( $\pm$ )-15–21 and olefination of the resultant aldehydes then gave the desired  $\alpha$ ,β-unsaturated esters ( $\pm$ )-22-28 in 18-57% overall yield (Scheme 1).

#### Scheme 1



When investigating  $PKR$ ,<sup>4</sup> we have promulgated that it is prudent to follow a strategy of first investigating the levels of substrate control offered by the chiral  $\alpha, \beta$ unsaturated ester upon conjugate addition of an achiral lithium amide, viz. lithium N-benzyl-N-isopropylamide 29. The levels of enantiorecognition between the chiral  $\alpha, \beta$ unsaturated ester (substrate) and lithium N-benzyl-N- $(\alpha$ methylbenzyl)amide 30 (chiral reagent) are then evaluated by investigation of their mutual kinetic resolution (MKR), i.e., addition of racemic lithium amide  $(\pm)$ -30 to racemic  $\alpha$ , $\beta$ -unsaturated ester. This approach eliminates any complicating effects of mass action and allows the maximum levels of enantiodiscrimination (as quantified by the factor,  $E^{6}$  to be very simply determined by analysis of the product distribution by <sup>1</sup>H NMR spectroscopy. Finally, having identified those substrates that undergo efficient MKR upon addition of racemic lithium amide  $(\pm)$ -30, their PKR employing a 50:50 pseudoenantiomeric mixture of enantiopure lithium  $N$ -benzyl- $N$ - $(\alpha$ -methylbenzyl) amide 30 and enantiopure lithium N-3,4-dimethoxybenzyl-N- $(\alpha$ -methylbenzyl)amide 31 may be performed. We therefore adopted this approach to investigate the potential of racemic γ-amino-α,β-unsaturated esters ( $\pm$ )-22–28 as substrates for our PKR protocol.

Addition of 1.6 equiv of lithium amide 29 to  $(\pm)$ -22-26 resulted in >95% conversion to the corresponding  $β, γ$ -diamino esters ( $\pm$ )-32-36 ( $\geq$ 88:12 dr in all cases), indicating high levels of substrate control. Chromatographic purification allowed isolation of  $\beta$ , *γ*-diamino esters ( $\pm$ )-32-36 in 60 $-86\%$  yield and in  $\geq$ 97:3 dr (Scheme 2). The relative 3,4syn-configuration within  $(\pm)$ -34 (R = Bn) was unambiguously established by single crystal X-ray diffraction analysis of the corresponding hydrochloride salt  $(\pm)$ -34•HCl. The relative 3,4-syn-configurations within  $(\pm)$ -32,  $(\pm)$ -33,  $(\pm)$ -35, and  $(\pm)$ -36 were assigned by analogy.<sup>7</sup> Meanwhile,  $(\pm)$ -28 proved recalcitrant to addition of lithium amide 29, even over extended reaction times and when the amount of lithium amide was increased from 1.6 to 10 equiv. Addition to  $(\pm)$ -27 proceeded with low levels of substrate control to give a 27:73 mixture of 3,4-syn-37:3,4-anti-43, which were isolated in 15 and 48% yield as single diastereoisomers (>99:1 dr). The relative 3,4 anti-configuration within 43 was unambiguously established by single crystal X-ray diffraction analysis, which therefore allowed the relative 3,4-syn-configuration within 37 to be assigned unambiguously (Scheme 2).



Previous investigations concerning conjugate addition of a range of nucleophiles to  $\alpha$ , $\beta$ -unsaturated carbonyl compounds with a stereocenter at the  $\gamma$ -position often invoke a modified Felkin-Anh model to rationalize the

<sup>(5)</sup> Reetz, M. T.; Röhrig, D. Angew. Chem., Int. Ed. Engl. 1989, 28, 1706.

<sup>(6)</sup> Horeau, A. Tetrahedron 1975, 31, 1307.

<sup>(7)</sup> Comparison with the substrate control elicited in the MKR of ( $\pm$ )-22-26 with lithium amide ( $\pm$ )-30, as well as in the PKR of ( $\pm$ )-22-26 with lithium amides  $(R)$ -30 and  $(S)$ -31, allows these configurational assignments to be made confidently. In addition, the relative configurations within  $\beta$ , $\gamma$ -diamino esters ( $\pm$ )-36 and ( $\pm$ )-42 [and (3S,4R, $\alpha$ R)-48] were subsequently unambiguously established by single crystal X-ray diffraction analysis of a cyclic derivative; see the Supporting Information for full experimental details.

<sup>(8)</sup> Yamamoto, Y.; Chounan, Y.; Nishii, S.; Ibuka, T.; Kitahara, H. J. Am. Chem. Soc. 1992, 114, 7652. Kireev, A. S.; Manpadi, M.; Kornienko, A. J. Org. Chem. 2006, 71, 2630.

observed diastereoselectivity.8 However, one other simplistic model (not dissimilar to a Felkin-Anh model) which is able to rationalize successfully the experimental data in this case uses insight obtained from single crystal X-ray diffraction analysis of  $\alpha$ , $\beta$ -unsaturated ester ( $\pm$ )-22. This revealed a solid state conformation in which the C(4) hydrogen atom lies almost perpendicular to the plane of the  $\alpha$ , $\beta$ -unsaturated system, with the bulky C(4)-N,N-dibenzylamino substituent occupying the less hindered "outside" position and the C(4)-methyl group in the more hindered "inside" position (Figure 1). Conjugate addition of lithium amide 29 to  $(\pm)$ -22 in this conformation would be predicted to occur on the least hindered face past the "small" hydrogen substituent to give  $(\pm)$ -3,4-syn-32, as observed experimentally. A similar analysis applied to  $\alpha$ , $\beta$ -unsaturated esters ( $\pm$ )-23-26 (R = Et, Bn, 'Bu, CH<sub>2</sub>OBn) would also successfully rationalize the observed substrate diastereofacial control, leading to  $(\pm)$ -3,4-syn-33-36. However, increased steric bulk of the C(4)-substituents in  $\alpha$ , $\beta$ -unsaturated esters ( $\pm$ )-27 and ( $\pm$ )-28 would serve to disfavor analogous conformations, thereby providing a rationale for their differing behavior. Presumably, the very large steric congestion around C(4) in the case of  $(\pm)$ -28  $(R = 'Pr)$  precludes addition of the sterically demanding lithium amide to C(3) completely.



Figure 1. Chem 3D representation of the single crystal X-ray diffraction structure of  $(\pm)$ -22 [(S)-enantiomer depicted; selected H atoms are omitted for clarity], and Newman projection along the  $C(3)-C(4)$  bond.

The extent of enantiorecognition between  $\alpha$ , $\beta$ -unsaturated esters  $(\pm)$ -22-26 (which offered high levels of substrate control) and lithium amide  $(\pm)$ -30 was next investigated, with high levels of enantiorecognition being expected. Indeed, addition of 1.6 equiv of lithium amide  $(\pm)$ -30 to  $(\pm)$ -22-26 gave, in each case, essentially a single diastereoisomeric product  $44-48$  in  $\geq 95:5$  dr, indicating very high levels of enantiorecognition between substrate and reagent, and consistent with  $E \ge 19^6$  in each case. Purification facilitated isolation of diastereoisomerically pure ( $> 99:1$  dr) samples of ( $\pm$ )-44-48. The relative  $(3RS, 4RS, \alpha SR)$ -configurations within ( $\pm$ )-45 and ( $\pm$ )-47 were unambiguously established by single crystal X-ray diffraction analyses, and therefore the relative configurations within  $(\pm)$ -44,  $(\pm)$ -46, and  $(\pm)$ -48 were assigned by analogy. It is notable that the relative configurations of the C(3)- and C( $\alpha$ )-stereogenic centers within both ( $\pm$ )-45 and

Scheme 3



 $a$  70% of a sample of 46 in 95:5 dr was also isolated.

 $(\pm)$ -47 [and, hence,  $(\pm)$ -44,  $(\pm)$ -46, and  $(\pm)$ -48] are in accordance with that predicted by the transition state mnemonic developed by us to rationalize the exceptional facial bias of this class of lithium amide.<sup>9</sup> This reagent control, when combined with that of the  $\alpha$ , $\beta$ -unsaturated ester (substrate control: production of the 3,4-syn-diastereoisomer favored), results in very highly selective reactions. These results suggest that  $\alpha$ , $\beta$ -unsaturated esters  $(\pm)$ -22-26 are viable substrates for our PKR protocol<sup>4</sup> (Scheme 3).

The PKR of  $\alpha$ , $\beta$ -unsaturated esters ( $\pm$ )-22-26 using a 50:50 pseudoenantiomeric mixture of lithium amides (R)- 30 (2 equiv) and  $(S)$ -31 (2 equiv) was next investigated. These reactions produced, in each case, a 50:50 mixture of the corresponding  $(\alpha R)$ -adducts 44-48 in  $\geq$ 95:5 dr and the  $(\alpha S)$ -adducts 49–53 in  $\geq$ 95:5 dr. Facile separation and purification via flash column chromatography allowed isolation of  $(\alpha R)$ -44-48 in >99:1 dr and 31-42% yield and  $(\alpha S)$ -49-53 in >99:1 dr and 40-43% yield. In each case, the product of addition of lithium amide  $(R)$ -30 was spectroscopically identical to the major diastereoisomer formed in the corresponding MKR reaction. Additionally, the relative configuration within  $\beta$ ,  $\gamma$ -diamino ester 49 was unambiguously established via single crystal X-ray diffraction analysis, with the absolute  $(3R, 4R, \alpha S)$ -configuration being assigned from the known  $(S)$ -configuration of the  $\alpha$ -methylbenzyl stereocenter. Given the pseudoenantiomeric nature of lithium amides  $(R)$ -30 and  $(S)$ -31, this analysis

<sup>(9)</sup> Costello, J. F.; Davies, S. G.; Ichihara, O. Tetrahedron: Asymmetry 1994, 5, 1999. For a review, see: Davies, S. G.; Smith, A. D.; Price, P. D. Tetrahedron: Asymmetry 2005, 16, 2833.

<sup>(10)</sup> Hoang, C. T.; Bouillère, F.; Johannesen, S.; Zulauf, A.; Panel, C.; Pouilhès, A.; Gori, D.; Alezra, V.; Kouklovsky, C. J. Org. Chem. 2009, 74, 4177.

<sup>(11)</sup> In the case of  $\beta$ ,  $\gamma$ -diamino ester **48**, the yield of the corresponding 4-aminopyrrolidin-2-one 68 (60% isolated yield) was somewhat compromised by the formation of methyl (3S,4R)-3,4-diacetamido-5-hydroxypentanoate 69 (25% isolated yield). This presumably arises from competing lactone formation (rather than lactam formation) from 58 under the reaction conditions, followed by methanolysis upon workup of the acetylation procedure.

### Scheme 4 Scheme 5



also allows the assigned relative  $(3RS, 4RS, \alpha SR)$ -configuration within racemic 44 to be unambiguously confirmed, with the absolute  $(3S, 4S, \alpha R)$ -configuration within enantiopure 44 following from the known  $(R)$ -configuration of the  $\alpha$ -methylbenzyl stereocenter. By similar reasoning, given the known relative configurations within racemic 45 and 47, the absolute  $(3S, 4S, \alpha R)$ -configurations within enantiopure 45 and 47 may be assigned from the known ( $R$ )-configuration of the  $\alpha$ -methylbenzyl stereocenter. Hence, the absolute  $(3R, 4R, \alpha S)$ -configurations within 50 and 52 can be unambiguously assigned. The absolute  $(3R, 4R, \alpha S)$ -configurations within 51 and 53 were assigned by analogy (Scheme 4).

With a range of enantiopure  $\beta$ ,  $\gamma$ -diamino esters in hand, their synthetic utility was demonstrated by elaboration of 44-48 to the corresponding 5-substituted 4-aminopyrrolidin-2-ones. Hydrogenolytic  $N$ -debenzylation of  $44-48$ was followed by acid-promoted cyclization $10$  to the corresponding 4-aminopyrrolidin-2-ones  $59-63$ , which were isolated as their acetate derivatives  $64-68$  in  $60-78\%$ yield over three steps. $11$  The absolute configurations within  $64-68$  were assigned from the known absolute



<sup>a</sup> Yield over 3 steps. <sup>b</sup>For **48**, R<sup> $\prime$ </sup> = Bn; for **58** and **63**, R<sup> $\prime$ </sup> = H; for **68**,  $R' = Ac$ .

configurations of the precursor  $\beta$ , *γ*-diamino esters **44–48**; <sup>1</sup>H NMR NOE analyses of  $64-68$  were also supportive of a relative 4,5-syn-configuration (Scheme 5).

In conclusion, conjugate addition of a 50:50 pseudoenantiomeric mixture of lithium  $(R)$ -N-benzyl-N- $(\alpha$ -methylbenzyl)amide and lithium (S)-N-3,4-dimethoxybenzyl- $N-(\alpha$ -methylbenzyl)amide to a range of racemic acyclic  $γ$ -amino-α, $β$ -unsaturated esters (derived from the corresponding  $\alpha$ -amino acids) effects their efficient parallel kinetic resolution, allowing the preparation of enantiopure β,γ-diamino esters. The β,γ-diamino ester products of these reactions are readily converted into the corresponding substituted 4-aminopyrrolidin-2-ones via N-debenzylation and cyclization. Further applications of this methodology are currently under investigation within our laboratory.

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Supporting Information Available. Experimental procedures, characterization data, copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra, and crystallographic information files (for structures CCDC 852571 $-852577$ ). This material is available free of charge via the Internet at http://pubs.acs.org.